



# Formula Grid

Write the correct formula that would be made if each of the following ions were crossed.

$\text{NH}_4^+$					
$\text{Al}^{3+}$					
$\text{Mg}^{2+}$					
$\text{Ca}^{2+}$					
$\text{Na}^+$					
Cation	$\text{Cl}^-$	$\text{S}^{2-}$	$\text{NO}_3^-$	$\text{SO}_4^{2-}$	$\text{OH}^-$
Anion					